



# Toxic Alcohol Ingestion:

An Overview of Diagnosis and  
Management

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## Disclosure Statement

- ▶ In accordance with ACPE standards, all relevant financial relationships with ineligible companies have been identified and mitigated. The faculty has no other conflicts of interest to disclose.

## Off-Label Discussion

- ▶ This activity may include discussion of unlabeled or unapproved uses of medications. All content is evidence-based and presented in a balanced, unbiased manner, drawing on established clinical guidelines and peer-reviewed data. Learners are encouraged to consult the official prescribing information for each product to review approved indications, contraindications, and warnings.

# Learning Objectives

At the conclusion of this CPE activity, participants should be able to:

1. **Recognize** the pathophysiology of methanol, isopropyl alcohol, and ethylene glycol poisoning
2. **Review** treatment options in patients presenting with a toxic alcohol ingestion
3. **Calculate** the serum osmolar gap and anion gap in a patient presenting with a toxic alcohol ingestion.

# Abbreviations used:

- ▶ CNS: Central nervous system
- ▶ N/V: Nausea/vomiting
- ▶ AG: anion gap
- ▶ AMS: altered mental status
- ▶ GI: gastrointestinal
- ▶ AACT: American Academy of Clinical Toxicology
- ▶ ADH: alcohol dehydrogenase
- ▶ ALDH: aldehyde dehydrogenase
- ▶ PPI: Proton pump inhibitor
- ▶ OG: osmolar gap
- ▶ AG: anion gap

# Background

- ▶ In the 2023 annual National Poison Data System report, 2.7% of total cases were exposed to alcohols
- ▶ Miscellaneous alcohols contributed to 7.6% of total exposures associated with fatalities
- ▶ Methanol, ethylene glycol, and isopropyl alcohol (isopropanol) are the most common toxic alcohols that are readily available and cheaper than alcohol
- ▶ The toxicity lies in their metabolites rather than the parent compound



<https://poisoncenters.org/national-poison-data-system>



# Diagnosis of Toxic Alcohol Ingestion

- ▶ Concentrations are measured by gas chromatography
  - ▶ Not readily available in hospitals
  - ▶ Only clinically useful within 2 hours of lab drawn
- ▶ Urine microscopy to detect calcium oxalate crystals (for ethylene glycol poisoning)
- ▶ Lab abnormalities (ABG, Chem7, **osmolality, anion gap**)
- ▶ Signs and symptoms

# Diagnosis (cont)

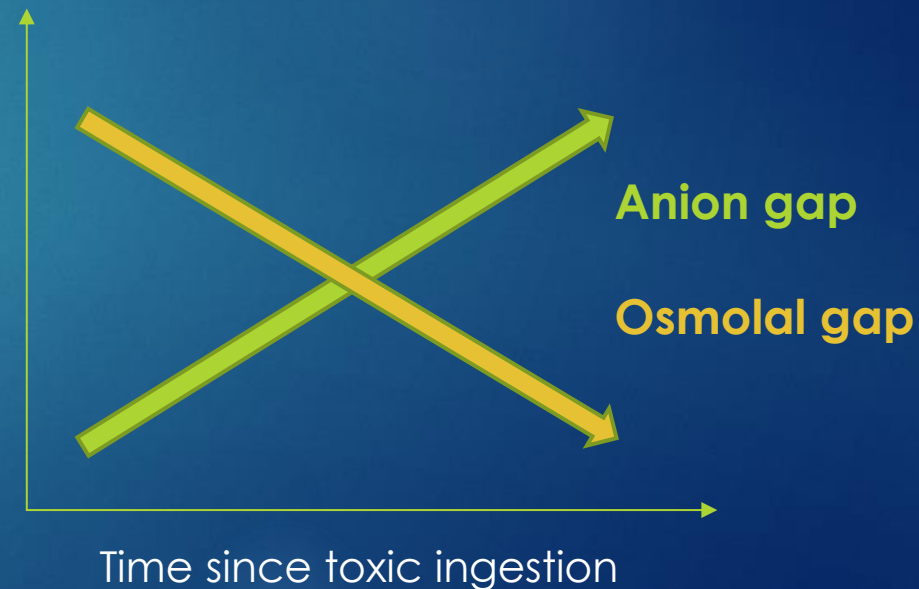
- ▶ In the absence of DKA or alcoholic ketoacidosis, the osmol and anion gap suggest methanol or ethylene glycol poisoning
  - ▶ Visual changes = Methanol
- ▶ Signs suggesting ethylene glycol toxicity
  - ▶ Inebriation without ethanol smell
  - ▶ Metabolic acidosis with large anion gap
  - ▶ Altered mental status
  - ▶ Osmolar gap
  - ▶ Calcium oxalate crystals with hypocalcemia
  - ▶ Renal injury

# Anion Gap(AG)

- ▶ **Anion Gap (mEq/L) = Na – (Cl + HCO<sub>3</sub>)**
  - ▶ Normal = 8-12 mEq/L
- ▶ Correct for hypoalbuminemia
- ▶ Albumin corrected AG (mEq/L) = anion gap + [ 2.5 × (4 – albumin (g/dL)) ]
- ▶ Elevated suggests metabolic acidosis
  
- ▶ Interpret in relation to time of ingestion! Early after ingestion, anion gap may be normal

# Osmolarity? Osmolality?

- ▶ Osmolarity: total number of particles in 1 L of solution (mOsm/L)
- ▶ Osmolality: total number of particles per kg (mOsm/kg)
- ▶ Osmolar and Osmolal are used interchangeably
  - ▶ The specific gravity of human serum= 1.01
  - ▶ 1 L of serum  $\approx$  1 kg
- ▶ Normal Serum Osm= 275-295 mmol/kg



# Osmolar Gap (OG)

- ▶ The osmolar gap (mOsm/L) is the difference between measured osmolality and calculated osmolarity

Calculated serum osmolarity (mOsm/L) =  $2 * Na + (BUN / 2.8) + (glucose / 18) + (ethanol / 4.6)$

*Na measured in mEq/L; BUN, glucose, and ethanol in mg/dL*

- ▶ Osmolar gap < 15-20 mOsm/L = ketoacidosis, lactic acidosis, renal failure
- ▶ Osmolar gap > 20 mOsm/L = suspicion for toxic alcohol poisoning
- ▶ OG doesn't reflect severity of poisoning, especially late in poisoning and absence of an OG doesn't exclude poisoning.

# Approach to the Osmolar Gap

1. **Serum osmolarity (mOsm/L)** =  $2 * \text{Na} + (\text{BUN} / 2.8) + (\text{glucose} / 18) + (\text{ethanol} / 4.6)$
2. Calculate the osmol gap (OG): **OG = measured Osm – calculated Osm**
3. You can then estimate toxic alcohol concentration with the osmolar gap:
4. **Toxic alcohol concentration = conversion factor \* OG**

Estimated alcohol concentration (mg/dL)	Conversion factor
Ethanol	4.6
Methanol	3.2
Ethylene glycol	6.2
Isopropyl alcohol	6.0



# The Three Most Common Toxic Alcohols

**METHANOL**

**ETHYLENE GLYCOL**

**ISOPROPANOL**

# Methanol (Methyl Alcohol)

- ▶ Colorless, volatile liquid with a faint sweet odor
- ▶ Commonly found in
  - ▶ Windshield-wiper fluid
  - ▶ Several types of antifreeze
  - ▶ Contaminated moonshine
  - ▶ Industrial solvents
- ▶ Least inebriating of the 3 alcohols
- ▶ First 24 hours: mild CNS depression, inebriation, visual disturbances
- ▶ Next 6-30 hours (Latent period): N/V, GI bleed, pancreatitis, abdominal pain, CNS depression, AMS, metabolic acidosis (delayed)
- ▶ Severe intoxication: tremors, muscle rigidity, basal ganglia hemorrhage



# Treatment of Methanol Poisoning

- ▶ Treatment
  - ▶ Prevent metabolism
    - ▶ Ethanol and fomepizole
  - ▶ Promote elimination
    - ▶ Hemodialysis
    - ▶ Folinic or folic acid (off-label use per FDA) until acidosis resolves
  - ▶ Supportive management
    - ▶ Treat acidosis with sodium bicarbonate at pH <7.3

# Treatment of Methanol Poisoning

- ▶ American Academy of Clinical Toxicology recommends fomepizole or ethanol to be given based on:
  - ▶ Plasma methanol concentration  $>20$  mg/dL, *or*
  - ▶ Recent history of ingestion of methanol with serum osmol gap  $>10$  mOsm/L, *or*
  - ▶ Strong clinical suspicion of poisoning with AT LEAST 2:
    1. Arterial pH  $<7.3$
    2. Serum  $\text{HCO}_3^- <20$  mEq/L
    3. Osmol gap  $>20$  mOsm/L

# Fomepizole (4-methylpyrazole, 4-MP)

- ▶ Competitive inhibitor of alcohol dehydrogenase (ADH)
- ▶ Preferred treatment for methanol AND ethylene glycol poisoning
- ▶ Does not cause CNS depression or hypoglycemia unlike ethanol
  - ▶ **Dose:** 15 mg/kg loading dose followed by 4 doses of 10mg/kg every 12 hours. After 48 hours: 15mg/kg every 12 hours.
- ▶ **Eliminated the need for dialysis in patients who don't have profound acidosis and no signs of optic or renal injury**

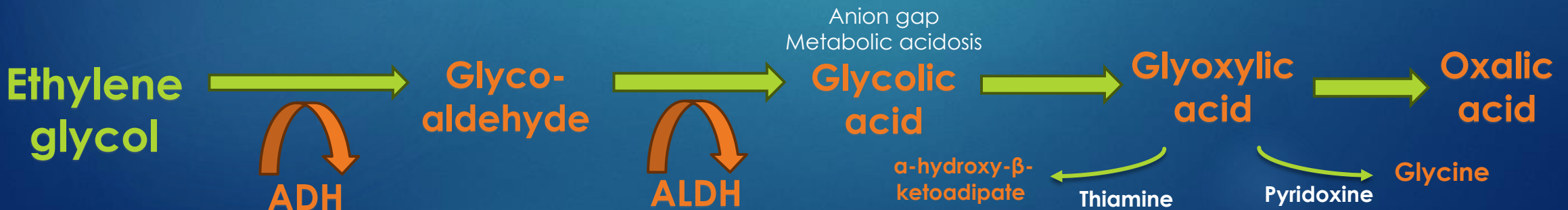


# Hemodialysis in Methanol Poisoning

- ▶ Extracorporeal treatments in poisonings workgroup (EXTRIP) developed recommendations for use of hemodialysis in severe methanol poisoning:
  1. Coma, seizures, new vision deficits
  2. Severe metabolic acidosis ( pH <7.15)
  3. Persistent metabolic acidosis despite adequate supportive measures/antidotes
  4. Anion gap > 24 mmol/L
  5. Serum methanol concentration > 50 mg/dL
  6. Renal failure
- ▶ **Endpoint:** undetected methanol concentration OR <25 mg/dL with normal acid-base balance

# Ethylene Glycol

- ▶ Colorless, sweet tasting fluid found in antifreeze
- ▶ 3 stages to poisoning:
  - ▶ 30 min-12hr after ingestion: anion gap metabolic acidosis, hematuria, neurological symptoms
  - ▶ 12-36 hrs: cardiopulmonary effects
  - ▶ 2-3 days: renal insufficiency/kidney failure
- ▶ Highest mortality in patients with severe acidosis (pH < 7.1) and longest time from exposure to treatment (>10 hr)



# Treatment of Ethylene Glycol Poisoning

- ▶ Similar to methanol poisoning
  - ▶ Blocking metabolism to its toxic byproducts
  - ▶ Ethanol or fomepizole
- ▶ AACT recommendations for treatment:
  1. EG level > 20 mg/dl or documented ingestion of toxic amounts
  2. Serum osmol gap > 10 mOsm/L
  3. OR clinical suspicion of EG poisoning and AT LEAST 2:
    - ▶ pH <7.3
    - ▶ bicarbonate concentration < 20 mE/L
    - ▶ presence of oxalate crystals in urine

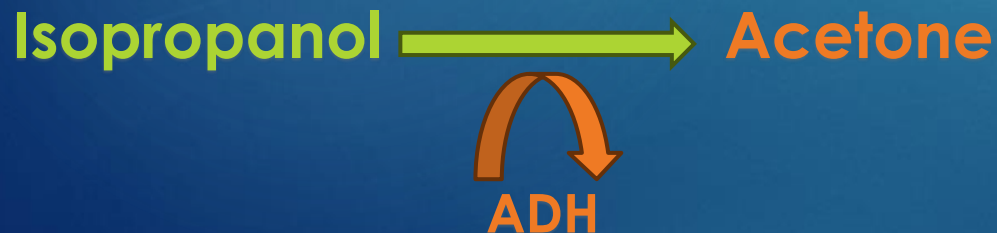
# Treatment of Ethylene Glycol Poisoning

- ▶ Cofactor administration: pyridoxine and thiamine
- ▶ Hemodialysis
- ▶ Per AACT when any of the clinical factors are present:
  - ▶ Persistent metabolic acidosis  $< 7.25$  pH despite therapy
  - ▶ Renal failure
  - ▶ EG levels  $> 50$  mg/dL
  - ▶ Glycolic acid levels 8-10 mmol/L

**Endpoint for dialysis:** Ethylene glycol levels  $< 20$  mg/dL and resolution of acid-balance

# Isopropyl Alcohol (Isopropanol)

- ▶ Primarily found in rubbing alcohol
- ▶ Symptoms similar to ethanol intoxication
  - ▶ GI, neurologic effects, pseudo-renal failure
  - ▶ Conversion to acetone gives off a fruity odor
  - ▶ Osmolar gap without acidosis
    - ▶ Normal acid-base status, hyper osmolality, and high acetone levels suggest poisoning
    - ▶ **Serum Acetone = (measured SCr – baseline Cr ) \* 100**



# Treatment of Isopropanol Poisoning

Supportive care!



- ▶ Airway management, IV hydration, and proton-pump inhibitors for GI bleed control
- ▶ Consider dialysis when:
  - ▶ Levels > 200 mg/dL and significant symptoms such as coma and/or hemodynamic instability
- ▶ Fomepizole is not indicated!
  - ▶ Inhibiting ADH prolongs isopropanol's half-life and symptoms

Putting  
them all  
together..

Alcohol	Methanol	Ethylene glycol	Isopropanol
Symptoms	Blurry vision GI, tremors/ CNS	CNS, cardiac, AKI/renal failure	GI, headache, coma
Lab abnormalities	↓ HCO <sub>3</sub>	↓Ca, ↓HCO <sub>3</sub>	Falsely ↑ SCr (normal BUN)
Osmolar gap	+ (early)	+ (early)	+ (early)
Anion gap	+ (late)	+ (late)	-
Serum acetone	-	-	+
Urine ketones	-	-	+
Urine oxalate crystals	-	+	-
Treatment	Ethanol, fomepizole, dialysis, folinic/folic acid	Ethanol, fomepizole, dialysis, pyridoxine, thiamine	Supportive, dialysis PPIs for GI bleed